

This is Science!

Name: _____ Date: _____ Group: _____

CLOZE-ING IN ON SCIENCE

INSTRUCTIONS: Use the words in the box below to fill in the blanks based on what you have learned about atoms. Words may be used more than once or not used at all.

Word Bank

4.00 protons	2 electrons	nucleus neutral	outside element	smallest neutrons	inside atom	largest
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CLOZE 1

Atoms are the smallest unit of an _____. Within the atom is a _____, which is the center of the atom. The _____ and neutrons are located within the _____. The _____ are outside the nucleus. The neutron has a _____ charge. A helium atom is the _____ unit of an element. The element helium has a mass of _____. A helium atom has _____ protons and _____ neutrons. The electrons are located on the _____ of the _____.

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Word Bank

protons	neutrons	nucleus	mass	atom	electron	small	electrical
atoms	large	small	electron	cloud	electrons	more	same

CLOZE 2

A _____ consists of the following subatomic particles known as _____ and _____. The _____ of the atom can be determined from the number of _____ and neutrons. _____ have a positive charge, and _____ have a negative charge. When an atom has the same number of protons and electrons, then the atom does not have a _____ charge. An atom also has _____ that are located on the outside of the _____. This area is known as the _____. The mass of the _____ is smaller than the mass of the protons and _____.